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## Erratum

# Erratum to: Kinetic measurement and prediction of the hydrogen outgassing from the polycrystalline LiH/Li<sub>2</sub>O/LiOH system by

L.N. Dinh et al., J. Nucl. Mater. 347 (2005) 31

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Eq. (3) of this paper needs the following clarification and correction.

The exact form of Eq. (3) should be:

$$\int_0^{\alpha} \ln\left(\frac{d\alpha}{dT/\beta}\right) d\alpha = -\frac{1}{R} \int_0^{\alpha} \frac{E d\alpha}{T} + \int_0^{\alpha} \ln(vf(\alpha)) d\alpha$$

If the activation energy barrier of the process under study is constant, the above equation can be simplified to Eq. (3) in the paper. Eq. (3) can be used as a test to see if the activation energy barrier of the process under study is constant or not. However, using Eq. (3) when  $E$  varies with  $\alpha$  (as done in this paper) is equivalent to approximating the value of  $E$  from 0 to  $\alpha$  with a constant  $E$ . A more accurate approach to numerically approximate values of  $E$  vs.  $\alpha$  is as following:

$$\begin{aligned} \int_0^{\alpha_i} \ln\left(\frac{d\alpha}{dt}\right) d\alpha + \int_{\alpha_i}^{\alpha_{i+1}} \ln\left(\frac{d\alpha}{dt}\right) d\alpha + \dots &= \left[ -\frac{E_i}{R} \int_0^{\alpha_i} \frac{d\alpha}{T} + \int_0^{\alpha_i} \ln(vf(\alpha)) d\alpha \right] \\ &+ \left[ -\frac{E_{i+1}}{R} \int_{\alpha_i}^{\alpha_{i+1}} \frac{d\alpha}{T} + \int_{\alpha_i}^{\alpha_{i+1}} \ln(vf(\alpha)) d\alpha \right] + \dots \end{aligned}$$

given the  $\alpha_i$  to  $\alpha_{i+1}$  interval is small.

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